Binary Fluorine Compounds

This table shows compounds of the element fluorine (F) with other elements:

— Radicals are not shown

— The table is not necessarily complete: Please report missing entries and better references to the given email address

— Compounds are arranged by the oxidation state of the non-fluorine element, or by stoichiometry

— Each compound is linked to its English Wikipedia page, or to another language Wikipedia page, or to a paper (in that order of preference)

— Not all compounds are equally stable; this table does not show which compounds are more stable

— Many compounds have multiple phases; only the most stable at standard conditions is indicated

— Many ionic and polymeric compounds have a molecular phase at high temperatues or low pressures; these are not indicated

Type of compound at standard conditions:

- Molecular with single non-fluorine atom (monomeric)
- Molecular with multiple non-fluorine atoms (oligomeric)
- Polymeric (1D/2D)
- Ionic or polymeric (3D)
- ***X*** Compounds in italics are not chemically stable at standard conditions
- Unknown or unclear structure
- X Predicted or contested
- Fluorine (for reference)
- [X] No binary compound with fluorine known
- Other elements: No information is available

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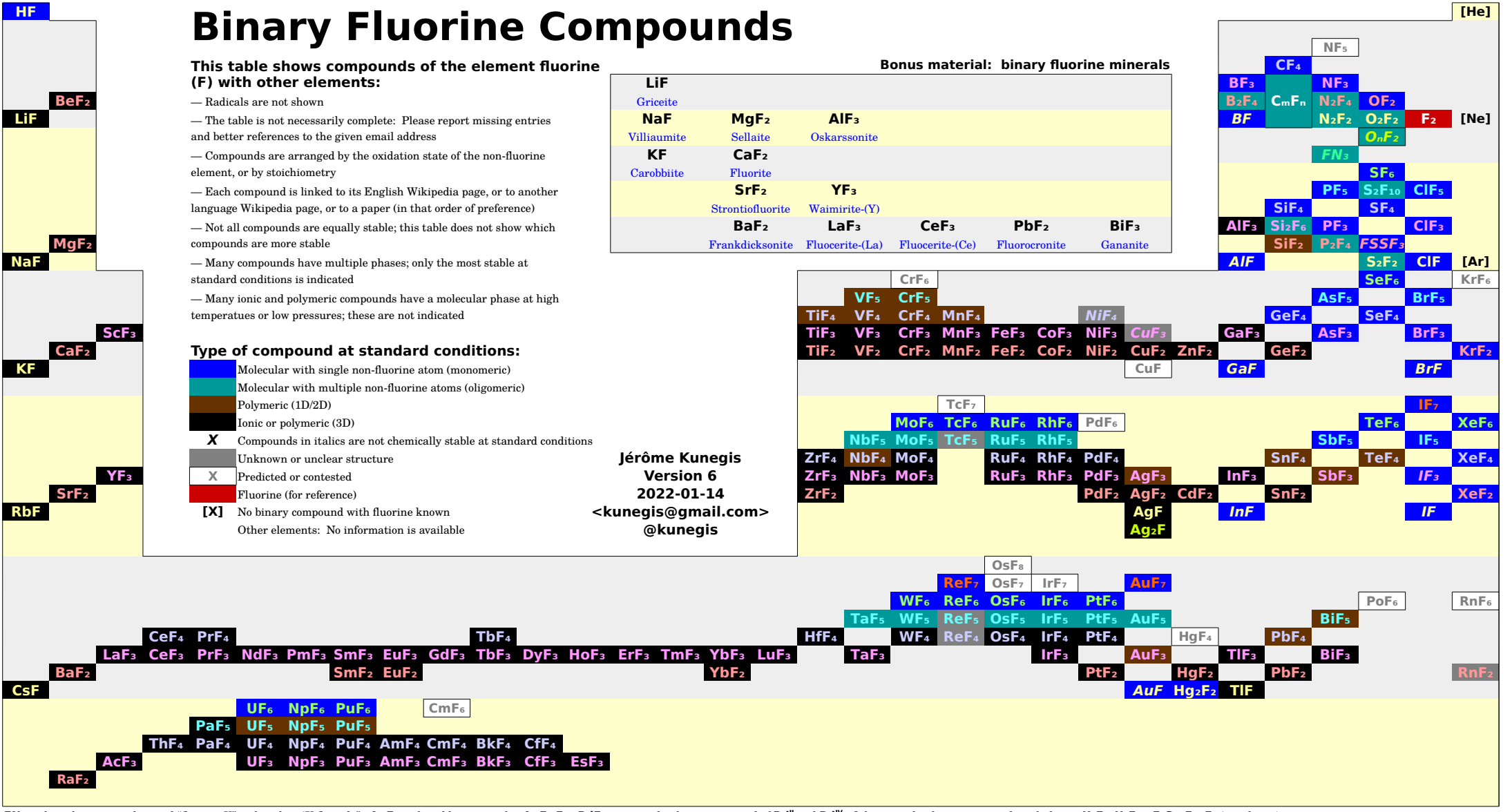
Bonus material: binary fluorine minerals

LiF					
Griceite					
NaF	MgF_2	AlF_3			
Villiaumite	Sellaite	Oskarssonite			
KF	CaF_2				
Carobbiite	Fluorite				
	SrF_2	YF_3			
	Strontiofluorite	Waimirite-(Y)			
	BaF_2	LaF_3	CeF_3	PbF_2	BiF_3
	Frankdicksonite	Fluocerite-(La)	Fluocerite-(Ce)	Fluorocronite	Gananite



FN_3 is the only compound named "fluorine X" rather than "X fluoride". **AuF_7** is the addition complex **$AuF_5 \cdot F_2$**. **PdF_3** is a mixed-valence compound of **Pd^{II}** and **Pd^{IV}**. Other mixed-valence compounds include e.g. **U_2F_9**, **U_4F_{17}**, **F_4S_3**, **Fe_2F_5** (not shown).